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## THE DECEPTIVE REARRANGEMENT OF A C8H9 CATION OF THE CAGE TYPE

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Summary. We produce evidence to show that the acetolysis of exo-seco-cubyl mesylate gives 7-tetracyclo  $[3.3.0.0^2, 80^3, 6]$  octyl acetate as sole product and not 5-tetracyclo  $[4.2.0.0^2, 40^3, 8]$ - octyl acetate.

We were surprised by the report that the acetolysis of the seco-cubyl mesylate <u>1</u>, proceeding through the intermediacy of the  $C_0H_9$  cation <u>2</u>, merely rearranged to the cyclopropyl-carbinyl cation <u>3</u> to yield the single epimeric acetate <u>4</u>.<sup>1</sup> A cation such as <u>2</u> would be expected to undergo a more deep-seated rearrangement.<sup>2</sup> In fact, we have recently obtained unequivocal proof that the Schmidt fragmentation reaction carried out on homocubanone <u>5</u> produces the rearranged cyano-mesylate 6.<sup>3</sup>

We believe that the latter reaction involves the initial formation of the imino-diazonium ion  $\underline{7}$  which cleaves to the cyano-*seco*-cubyl cation 8 which further rearranges to the cyclo-propylcyclobutylcarbinyl cation  $\underline{9}$ , to be finally captured by solvent, thereby affording the mesylate 6.

Doubting that two carbocations so similar as  $\underline{2}$  and  $\underline{8}$  would evolve differently, we converted the mesylate  $\underline{6}$  into the acetate  $\underline{10}$  and compared its  ${}^{13}$ C-NMR spectra with that of the compound purported to have structure  $\underline{4}$ . The similarity of the spectra was striking (Table). In order to confirm that the acetate which was assigned structure  $\underline{4}$  had in reality structure  $\underline{11}$ , we decided to prepare, in unequivocal manner, the acetate 11 and its alcohol 12.

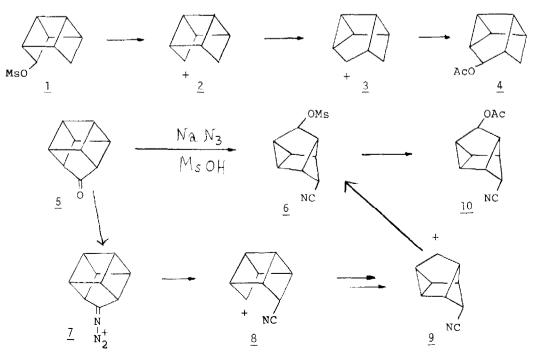
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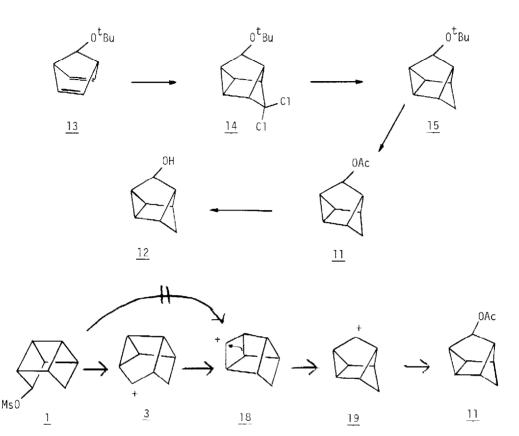
			9 <sup>10</sup> 0-C0-CH <sub>3</sub> 0-C(CH <sub>3</sub> ) <sub>3</sub> 0H0-C0-CH <sub>3</sub>				
	F						
	F <u>16</u>	cí <u>17</u>	NC 10	<u>15</u>	<u>12</u>	<u>11</u> H <sub>3</sub> C	$-COO = 4^{\overline{b}}$
C1]	12.91	18.95	18.95 D	19.31 D	19.83 D	20.03 D	C2 20.0 D
C2 🕇	12.91	18.95	19.68 D	20.63 D	20,53 D	20.49 D	C3 20.5 D
C3]	<b>49.</b> 19	58.20	41.85 D	38,99 D	39.66 D	39.40 D	C8 39.4 D
C5 🕇	49.19	58.20	43.61 D	40.98 D	40.03 D	40.97 D	C1 41.0 D
C4	128.10	91.69	44.29 D	43.45 T	43.10 T	43.30 T	C7 43.4 T
C6	33.81	40.62	45.80 D	48.96 D	49.13 D	46.23 D	C6 46.2 D
С7	31.77	34.38	79.82 D	78.60 D	78.54 D	81.16 D	C5 81.3 D
C8	25.61	30.65	34.45 D	34.37 DD	35.13 DD	32.03 DD	C10 32.0 Q
С9	-	-	171.30 S	72.83 S	-	170.94 (S)	C9 171.3 S
C10	-	-	20.90 (Q)	28.67 Q	-	21.22 Q	C4 21.2 D

Table. <sup>13</sup>C-Resonances of Some Derivatives of Tetracyclo[3.3.0.0<sup>2,8</sup>0<sup>3,6</sup>]octane.<sup>*a*,*c*</sup>

<sup>a</sup>Spectra recorded at 25.2 MHz in CDCl<sub>3</sub> solution; values with respect to tetramethylsilane. <sup>b</sup>Structure proposed and resonances assigned thereto according to reference 1.

 $^{c}$ Off-resonance multiplicities are indicated by D = doublet, etc.





Exploitation of the homo-1,4 cheletropic reaction<sup>4</sup> of dichlorocarbene with 7-t-butoxynorbornadiene <u>13</u> gave the adduct <u>14</u>. Appropriate reductive dechlorination of the *gem*-dichloro function under the Winstein conditions<sup>5</sup> preserved the skeleton<sup>6</sup> and yielded the ether of the parent hydrocarbon <u>15</u>. Acid hydrolysis and acetylation gave a mixture of products. Preparative gas-liquid chromatography permitted the isolation of the desired acetate <u>11</u> in 40% yield. Reduction of <u>11</u> with lithium aluminium hydride gave the corresponding alcohol <u>12</u>.

Comparison of the <sup>13</sup>C-NMR spectra of <u>10</u>, <u>11</u>, <u>12</u> and <u>15</u> with those of the difluoro and dichloro carbene adducts obtained from norbornadiene (<u>16</u> and <u>17</u>) and that reported for <u>4</u> reveals, without a shadow of doubt, that <u>4</u> in fact has the structure <u>11</u>.

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The spectral arguments may be briefly resumed as follows. (2) <sup>1</sup>H-NMR spectra of our compounds <u>11</u> and <u>12</u> correspond exactly to those described for the acetate and alcohol of the supposed structure <u>4</u>. (2) The <sup>13</sup>C resonances of our structure <u>11</u> are identical with those of the compound assigned structure <u>4</u>. (3) The off-resonance "quadruplet" of the signal attributed to Cl0 in <u>4</u> is really a doublet of doublets. On the other hand, the signal at 21.2 ppm is a quartet, in keeping with its assignment to C9 in our structure <u>11</u>. (4) Conversion of the acetate to the alcohol (<u>11</u> to <u>12</u>) is most revealing; the two resonances assigned to C4 and C9 in <u>4</u> disappear, whereas the rest remain essentially unchanged, especially for those atoms furthest from the modified function. Consequently, the C4 signal has been mis-assigned.

A last question concerns the pathway by which the acetolysis of <u>I</u> leads uniquely to <u>11</u>. The most economical process is the concerted Wagner-Meerwein shift which generates the cyclobutyl cation <u>18</u> directly. Subsequent ring contraction, followed by attack by solvent acetic acid on either face of the cationic center of <u>19</u> give the product <u>11</u>. However, calculation reveals<sup>7</sup> and trapping experiments<sup>8</sup> have confirmed that the rearrangement occurs via the Dauben cation <u>3</u> which then gives 18.

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- 6) The concordance of the <sup>13</sup>C-NMR shifts of <u>15</u> with those of <u>17</u> and <u>10</u> secures its structure. The structure of the 7-bromo derivative <u>10</u> has been determined by radio-crystallography (T.S. Cameron, unpublished work).
- 7) The heats of formation and minimum energy reaction paths of the *seco*-cubyl cation and its congeners are being calculated by the MINDO/3 method (J.A. Zuber, unpublished work).
- 8) G. Mehta & S.C. Suri, unpublished results.

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